QUANTUM PHYSICS-III

Time: Three Hours] Note: Attempt five questions in all. Select one question each from Section A, B, C, and D Q. No. 9 Section E is compulsory, attempt any five parts from it. Use of Non-programmable calculator is allowed.		
Section: A		
1.	(a) (b)	Derive Uncertainty principle, and give its two main applications. 6 Find the ratio between the wavelengths of the de-Broglie waves for electrons in the second and third Bohr's orbits in hydrogen atom.
2.		Explain the meaning of well-behaved wave function. 2 the prove Ehrenfest theorem.
3.	(0)	Section: B
٥.	(a)	What is Harmonic oscillator? Solve the Schrodinger equation for one-dimentional harmonic oscillator and obtain energy levels.
	(b)	Derive Time-independent Schrodinger wave equation.
4.	Con	sider a particle is incident on a potential step of hight V0 with energy E > V0. Calculate the
	refle	ection and transmission coefficient.
		Section : C
5.	(a)	Describe Frank-Hertz experiment. What does it demonstrate 2
	(b)	What die Stern-Gerlach experiment establish? The the radial part of he Schrodinger wave equation for hydrogen atom to othain its energy list. Section 10 Sectio
6.	Solv	the radial part of he Schrodinger wave equation for hydrogen atom to other its energy
	ieve	is.
7.	(a`	Section : D
٠.	(a	Distinguish between continuous and characteristic spectrum of X-rays, and explain their origin.
	(b)	William I and the second of th
	(c)	How the quantity and quality of X-rays can be controlled in a X-ray tube? Why do all molecules not show rotational spectra? Explain the origin of Stokes and Anti-stokes lines in Raman effect on quantum theory. Section: F
	(d)	How the quantity and quality of X-rays can be controlled in a X-ray tube? Why do all molecules not show rotational spectra?
8.	(a)	Explain the origin of Stokes and Anti-stokes lines in D
	(b)	Explain the origin of Stokes and Anti-stokes lines in Raman effect on quantum theory. 3 State and explain Mosley's law.
		Section: E
Compular		
9.	Exp	alin any five of the following in short:
	(4)	Wildlight Dialomic and Polyatomic males 1 a a a
	(b)	What are the various possible motions in a molecule? What is meant by Normalization of a molecule?
	(c)	What is meant by Normalization of a wave function? Discuss.
	(e)	What are Orthogonal wave function? Discuss.
	(0)	coordinates 2 atom, why the solution of Schrodinger wave equation:
	(f)	For hydrogen atom, why the solution of Schrodinger wave equation involves spherical polar What is normal Zeeman effect and anomals.
		What is the condition for most and allomalous Zeeman effect 2
	(g) (h)	What is Tunnel effect? How does it explain Alpha-decay?
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		does it explain Alpha-decay?