

STATISTICAL PHYSICS AND THERMODYNAMICS - I

Time : Three Hours]

Note : Attempt *one* question each from Sections A, B, C, and D carrying 15 marks each, and *five* questions from Section E consisting of *eight* short answer type questions carrying 3 marks each.

[Maximum Marks : 75

- Section : A
1. (a) Taking the case of n particle distributed in two compartments, discuss the variation of probability of a macrostate on account of small deviation from the state of maximum probability. 10
 - (b) In a system of 8-distinguishable particle, distribution is in two equal size compartments. Calculate the probability of the macrostate (2, 6). 5

2. (a) Discuss the distribution n -distinguishable particle in k compartment of unequal size which are further sub-divided into cells of equal a priori probability. 10
 (b) Prove that for dynamic system, the fraction of total time spent in any particular microstate is proportional to the thermodynamical probability of that state. 5
- Section : B**
3. Treating an ideal gas as a system governed by classical statistics, derive the Maxwell-Boltzmann law of distribution of molecular speed. 15
4. (a) Discuss Planck's law for the energy distribution of the black body. 10
 (b) Find the number of phase-space cells in the energy interval E and $E + dE$. 5
- Section : C**
5. (a) Derive an expression for entropy of a perfect gas. 10
 (b) Show that the entropy of a thermodynamic system remains constant in any reversible process. 5
6. (a) Explain in brief Carnot cycle and S - T diagram for Carnot cycle. What is the efficiency of Carnot's heat engine. 3+3+4
 (b) Discuss the heat death of universe. 5
- Section : D**
7. (a) Show that $C_p - C_v = -TE\alpha^2V$, where T is temperature, E = Bulk modulus of elasticity, α = Coefficient of expansion and V = Specific volume. 10
 (b) Calculate the temperature of inversion for a real gas whose constant $a = 0.0245 \text{ m}^4 \text{ Nmole}^{-2}$, $b = 2.67 \times 10^{-5} \text{ m}^3 \text{ mole}^{-1}$ and $R = 8.4 \text{ J mol}^{-1} \text{ K}^{-1}$. 5
8. (a) Explain Joule Thomson effect. Derive expression of its coefficients. 10
 (b) Calculate the Coefficient of performance of a refrigerator operating between the temperature 5°C and 35°C . 5
- Section : E**
9. Attempt any five questions in short of the following :
- (i) Eight similar coins are tossed for a large no. of times. Calculate probability of getting the heads of 5 coins uppermost. 3
 (ii) What is the role of Elementary cell in Statistical Physics ? 3
 (iii) Calculate the r.m.s. velocity and most probable velocity of nitrogen at 0°C . Given $K = 13.8 \times 10^{23} \text{ J/K}$. 3
 (iv) What do you understand by Fermi energy level of metal ? 3
 (v) What are Reversible and Irreversible processes ? 3
 (vi) What does the internal energy of an ideal gas differ from that of real gas ? 3
 (vii) Derive relation between Adiabatic and Isothermal elasticity. 3
 (viii) Write four thermodynamical relations. 3